ASYMMETRIC SYNTHESIS OF 2-EXO-AMINONORBORNANE-2-ENDO-CARBOXYLIC ACID DERIVATIVES

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Abstract: The Diels-Alder reaction of cyclopentadiene with N-acetyl- α , β -didehydroalaninate of (-)-cis-3-hydroxy isobornyl neopentyl ether yields the corresponding 2-endo(exo)-acetamido-5-norbornene-2-exo(endo)-carboxylates with total diastereofacial selectivity and with a preference for the cycloadduct with the ester group placed at the endo position. So this dienophile is complementary with the previously described (-)-menthyl N-acetyl- α , β -didehydroalaninate for the asymmetric synthesis of 2-aminonorbornane-2-carboxylic acids, since with the latter dienophile exo cycloadducts are preferably obtained with a high diastereofacial selectivity.

Cycloaliphatic amino acids possessing a norbornane skeleton are of interest because of their biological properties¹ and they have been obtained by Diels-Alder reaction of N-acyl- α , β -didehydroalaninates and cyclopentadiene². We have recently described the first asymmetric synthesis of 2-aminonorbornane-2-carboxylic acids using (-)-menthyl N-acetyl- α , β -didehydroalaninate (<u>3a</u>) as a dienophile³. The Diels-Alder reaction of cyclopentadiene with this dienophile (Scheme 2) leads preferably to exo cycloadducts (<u>6a</u> + <u>7a</u>) with high diastereofacial selectivity. However, endo cycloadducts (<u>4a</u> + <u>5a</u>) are obtained in a minor proportion and with only a moderate diastereofacial selectivity. Therefore, this reaction is an excellent method for the asymmetric synthesis of 2-endo-aminonorbornane-2-exo-carboxylic acids, but it is not very efficient in the preparation of 2-exo-aminonorbornane-2-endo-carboxylic acids. In this paper we want to describe the use of the N-acetyl- α , β -didehydroalaninate of (-)-cis-3-hydroxy isobornyl neopentyl ether (<u>3b</u>) as a dienophile that leads mainly to endo cycloadducts (<u>4b</u> + <u>5b</u>) with total diastereofacial selectivity.

RESULTS AND DISCUSSION

The chiral dienophile (3b), obtained by the reaction of methyl N-acetyl- α , β -didehydroalaninate (1)^{2b,4} with (-)-cis-3-hydroxy isobornyl neopentyl ether (2b)⁵ in the presence of trimethylaluminium⁶, was made to react with cyclopentadiene under several conditions. The results obtained from these Diels-Alder reactions were determined by ¹H-NMR. In order to assign the NMR signals to the corresponding cycloadducts, these were prepared, as endo (4b + 5b) and exo (6b + 7b) mixtures, by reaction of the corresponding endo (12 + 13) and exo (14 + 15) methyl 2-acetamido-5-norbornen-2-carboxylates^{2b} with the chiral auxiliary (2b) in the presence of trimethylaluminium, and analyzed by ¹H-NMR. Scheme 1 shows the signals that can be used to determine the relative amounts of dienophile and cycloadducts in the mixture. As can be seen the % conversion, the endo/exo and the endo/endo (4b : 5b) ratios can easily be determined by the integration of the acetamido singlets, but the determination of the exo/exo (6b : 7b) ratio is not so straightforward. Luckily the ¹H-NMR spectra obtained from the reaction mixtures were simpler than expected. So, only one endo cycloadduct was observed in the acetamido region (5b, 1.95 ppm), and the H_α region showed signals at 4.75 and 4.80 ppm. Furthermore, the ratios between the peaks at 1.95 and 1.86 ppm (endo/exo) and that of the doublets at 4.75 and 4.80 ppm were the same, showing that the signal at 4.75 ppm corresponded to <u>5b</u>, without contamination of <u>7b</u>, and <u>6b</u> was the only exo cycloadduct present in the mixture.



Scheme 1

In order to confirm these determinations and to determine the absolute configuration of the cycloadducts obtained, the following experiment was carried out. In a previous paper we have described how the reaction between cyclopentadiene and (-)-menthyl N-acetyl- α , β -didehydroalaninate³ leads preferably to 5a and 6a. A mixture coming from this Diels-Alder reaction was separated into its endo and exo components by column chromatography (diethyl ether : n-hexane = 9:1 as an eluent), and the purified cycloadducts were almost quantitatively transformed into the corresponding methyl 2-acetamido-5-norbornene-2-carboxylates (12 + 13 and 14 + 15) by saponification and methylation with diazomethane. These mixtures were analyzed by ¹H-NMR in

the presence of $Eu(tfc)_3$ (L:S molar relationship = 0.85, CD₃CN as solvent) and the singlets corresponding to the acetamido and ester methyls were assigned to each enantiomer (Scheme 2).



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Scheme 2

The mixtures of cycloadducts obtained from the Diels-Alder reactions of <u>3b</u> with cyclopentadiene were separated into the endo (<u>4b</u> + <u>5b</u>) and exo (<u>6b</u> + <u>7b</u>) components by means of column chromatography (CHCl₃ : diethyl ether = 7: 3 as an eluent). The purified cycloadducts were transformed into the corresponding methyl 2acetamido-5-norbornene-2-carboxylates (<u>12</u> + <u>13</u> and <u>14</u> + <u>15</u>) by the above described procedure and they were analyzed by ¹H-NMR in the presence of Eu(tfc)₃ (L:S molar relationship = 0.85, CD₃CN as solvent) showing that only one enantiomer is present in the mixture. In order to ensure the position of the signals, a known amount of racemic <u>12</u> + <u>13</u> (or <u>14</u> + <u>15</u>) was added and the ¹H-NMR analysis was repeated. In this way it was determined that the Diels-Alder reaction of <u>3b</u> with cyclopentadiene led to <u>5b</u> and <u>6b</u>.

Table 1 gathers the results obtained from the reaction between $\underline{3b}$ and cyclopentadiene, together with some comparative results from the reaction between $\underline{3a}$ and cyclopentadiene.

<u>Table 1</u> .	$Diels-Alder\ reaction\ between\ N-acetyl-\alpha,\beta-didehydroalaninate\ of\ (-)-cis-3-hydroxy\ isobornyl$
	neopentyl ether (3b) and cyclopentadiene

TiCl4 (0.75) $3a^c$ 25 3 100 $19/81$ $86/14$ $97/3$ TiCl4 (0.75) $3b$ 25 4 95 $55/45$ $>99/1$ $>99/1$ TiCl4 (1.1) $3a^c$ 0 3 100 $21/79$ $86/14$ $92/8$ TiCl4 (1.1) $3b$ 0 1.5 94 $62/38$ $>99/1$ $>99/1$ TiCl4 (0.5) $3b$ 0 4 100 $62/38$ $>99/1$ $>99/1$ TiCl4 (0.3) $3b$ 0 21 7.5 $ -$ TiCl4 (0.75) $3b$ -70 9 60 $70/30$ $>99/1$ $>99/1$ AlCl3 (1.1) $3a^c$ 25 22 74 $26/74$ $81/19$ $90/10$ AlCl3 (0.75) $3b$ 25 72 0 $ -$ AlCl3 (1.1) $3b$ 445 72 20 $55/45$ $>99/1$ $>99/1$	Lewis acid (eq)	Dienophile	<u>T°C</u>	<u>t h</u>	<u>% Yield</u> a	ratio (4+5)/(6+7) ^a	ratio 5/4 ^{a,b}	<u>ratio 6/7</u> a,b
TiCl4 (0.75) $3b$ 25 4 95 $55/45$ $>99/1$ $>99/1$ TiCl4 (1.1) $3a^c$ 0 3 100 $21/79$ $86/14$ $92/8$ TiCl4 (1.1) $3b$ 0 1.5 94 $62/38$ $>99/1$ $>99/1$ TiCl4 (0.5) $3b$ 0 4 100 $62/38$ $>99/1$ $>99/1$ TiCl4 (0.3) $3b$ 0 21 7.5 $ -$ TiCl4 (0.75) $3b$ -70 9 60 $70/30$ $>99/1$ $>99/1$ AlCl3 (1.1) $3a^c$ 25 22 74 $26/74$ $81/19$ $90/10$ AlCl3 (0.75) $3b$ 25 72 0 $ -$ AlCl3 (1.1) $3b$ 445 72 20 $55/45$ $>99/1$ $>99/1$	TiCl4 (0.75)	3a ^c	25	3	100	19/81	86/14	97/3
TiCl4 (1.1) $3a^c$ 03100 $21/79$ $86/14$ $92/8$ TiCl4 (1.1) $3b$ 01.5 94 $62/38$ $>99/1$ $>99/1$ TiCl4 (0.5) $3b$ 04100 $62/38$ $>99/1$ $>99/1$ TiCl4 (0.3) $3b$ 0 21 7.5 TiCl4 (0.75) $3b$ -709 60 $70/30$ $>99/1$ $>99/1$ AlCl3 (1.1) $3a^c$ 25 22 74 $26/74$ $81/19$ $90/10$ AlCl3 (0.75) $3b$ 25 72 0 AlCl3 (1.1) $3b$ 25 72 0 AlCl3 (1.1) $3b$ -45 72 20 $55/45$ $>99/1$ $>99/1$	TiCl4 (0.75)	3b	25	4	95	55/45	>99/1	>99/1
TiCl4 (1.1) $3b$ 01.594 $62/38$ >99/1>99/1TiCl4 (0.5) $3b$ 04100 $62/38$ >99/1>99/1TiCl4 (0.3) $3b$ 0217.5TiCl4 (0.75) $3b$ -709 60 70/30>99/1>99/1AlCl3 (1.1) $3a^c$ 252274 $26/74$ $81/19$ 90/10AlCl3 (1.1) $3b$ 2572 34 $44/56$ >99/1>99/1AlCl3 (0.75) $3b$ 25720AlCl3 (1.1) $3b$ -457220 $55/45$ >99/1>99/1	TiCl ₄ (1.1)	3ac	0	3	100	21/79	86/14	92/8
TiCl4 (0.5) $3b$ 04100 $62/38$ >99/1>99/1TiCl4 (0.3) $3b$ 021 7.5 TiCl4 (0.75) $3b$ -709 60 $70/30$ >99/1>99/1AlCl3 (1.1) $3a^c$ 2522 74 $26/74$ $81/19$ 90/10AlCl3 (1.1) $3b$ 25 72 34 $44/56$ >99/1>99/1AlCl3 (0.75) $3b$ 25 72 0AlCl3 (1.1) $3b$ -45 72 20 $55/45$ >99/1>99/1	TiCl ₄ (1.1)	3ь	0	1.5	94	62/38	>99/1	>99/1
TiCl4 (0.3) $3b$ 0217.5TiCl4 (0.75) $3b$ -709 60 $70/30$ >99/1>99/1AlCl3 (1.1) $3a^c$ 252274 $26/74$ $81/19$ 90/10AlCl3 (1.1) $3b$ 2572 34 $44/56$ >99/1>99/1AlCl3 (0.75) $3b$ 25720AlCl3 (1.1) $3b$ -45722055/45>99/1>99/1	TiCl ₄ (0.5)	3ь	0	4	100	62/38	>99/1	>99/1
TiCl4 (0.75) $3b$ -70 9 60 $70/30$ $>99/1$ $>99/1$ AlCl3 (1.1) $3a^c$ 25 22 74 $26/74$ $81/19$ $90/10$ AlCl3 (1.1) $3b$ 25 72 34 $44/56$ $>99/1$ $>99/1$ AlCl3 (0.75) $3b$ 25 72 0 $ -$ AlCl3 (1.1) $3b$ -45 72 20 $55/45$ $>99/1$ $>99/1$	TiCl4 (0.3)	3b	0	21	7.5	-	-	-
AlCl3 (1.1) $3a^c$ 25 22 74 $26/74$ $81/19$ $90/10$ AlCl3 (1.1) $3b$ 25 72 34 $44/56$ $>99/1$ $>99/1$ AlCl3 (0.75) $3b$ 25 72 0 $ -$ AlCl3 (1.1) $3b$ -45 72 20 $55/45$ $>99/1$ $>99/1$	TiCl4 (0.75)	3ъ	-70	9	60	70/30	>99/1	>99/1
AlCl3 (1.1) 3b 25 72 34 44/56 >99/1 >99/1 AlCl3 (0.75) 3b 25 72 0 - - - AlCl3 (1.1) 3b -45 72 20 55/45 >99/1 >99/1	AlCl ₃ (1.1)	3ac	25	22	74	26/74	81/19	90/10
AlCl3 (0.75) 3b 25 72 0 - - AlCl3 (1.1) 3b -45 72 20 55/45 >99/1 >99/1	AlCl ₃ (1.1)	3Ъ	25	72	34	44/56	>99/1	>99/1
AlCl ₃ (1.1) 3b -45 72 20 55/45 >99/1 >99/1	AlCl ₃ (0.75)	3ь	25	72	0	-	-	-
	AlCl ₃ (1.1)	3ь	-45	72	20	55/45	>99/1	>99/1

a. Determined by ¹H-NMR

b. No signals corresponding to <u>4b</u> and <u>7b</u> could be found.

^c. Ref. 3

The rate of the Diels-Alder reaction depends on the Lewis acid used as a catalyst, and is faster with TiCl4. With AlCl3 an equimolecular amount of catalyst is needed to achieve low to moderate yields, suggesting a low constant of formation of the dienophile-AlCl3 complex. The reaction of (-)-menthyl N-acetyl- α , β -didehydroalaninate (3a) with AlCl3 is faster, but a similar dependence on the nature of the catalyst is observed³.

Table 1 shows that, starting from <u>3a</u>, <u>5a</u> is obtained with an overall yield of about 20%, whereas the yield in <u>6a</u> is about 70%. On the other hand, <u>3b</u> leads to <u>5b</u> with a yield of 62% and to <u>6b</u> with a yield of 38%. Given that the cycloadducts can be easily transformed into the corresponding 2-aminonorbornane-2-carboxylic acids³, both dienophiles are complementary for the asymmetric synthesis of the α -amino acids with the amino group placed at the endo or exo position.

Although, in Lewis acid-catalyzed reactions with cyclopentadiene the acrylate of (-)-cis-3-hydroxy isobornyl neopentyl ether⁵ leads to better endo/exo ratios than (-)-menthyl acrylate⁷, we cannot suggest an explanation for the different endo/exo selectivity obtained with <u>3a</u> and <u>3b</u>, and in particular for the unexpected behaviour of <u>3b</u>.

In the Diels-Alder reaction between the acrylate of (-)-cis-3-hydroxy isobornyl neopentyl ether and cyclopentadiene, (1R,2S,4R)-5-norbornene-2-carboxylate is preferably obtained with high diastereofacial selectivity⁷. The stereochemical control may be explained by the similarity to the model proposed for classical acrylates⁸ where the carbonyl group is antiplanar to the olefinic double bond and synplanar to the alcoxy C-H group (Figure 1). Accordingly, the Re-face of the dienophile is shielded by the neopentyl group of the chiral auxiliary and the attack of the diene takes place on the Si-face leading to <u>5b</u> and <u>6b</u>. If a preferable co-ordination of the Lewis acid to the acetamido group is admitted, the anti enoate conformer may be favoured by the formation of an intramolecular hydrogen bond (Figure 1).



Figure 1

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EXPERIMENTAL SECTION

- * Methyl N-acetyl- α , β -didehydroalaninate (1)
- * (-)-menthyl N-acetyl-α,β-didehydroalaninate (3a)
- * N-acetyl-α,β-didehydroalaninate of (-)-cis-3-hydroxy isobornyl neopentyl ether (<u>3b</u>) These compounds were prepared following the methods described in the literature^{2b,6}
- * (1R,2S,4R)- 2-exo-acetamidobicyclo[2.2.1]hept-5-ene-2-endo-carboxylate (<u>5b</u>) and (1S,2S,4S)-2-endo-acetamidobicyclo[2.2.1]hept-5-ene-2-exo-carboxylate (<u>6b</u>) of (-)-cis-3hydroxy isobornyl neopentyl ether.

General procedure for Diels-Alder reactions: To a solution of N-acetyl- α_{β} -didehydroalaninate of (-)-cis-3-hydroxy isobornyl neopentyl ether (<u>3b</u>) (175.5 mg, 0.5 mmol) in anhydrous toluene (20 ml) under an inert atmosphere the catalyst was added. After 20 minutes stirring at the corresponding working temperature, a solution of freshly distilled cyclopentadiene (132 mg, 2 mmol) in anhydrous toluene (0.5 ml) was added and the reaction was stirred for the time reported in Table 1. The mixture was treated with Na₂CO₃.10H₂O (200 mg), the solution was filtered and the solvent evaporated under vacuum to give a mixture, the composition of which was analyzed by ¹H-NMR. The reaction mixture was separated into endo (<u>5b</u>) and exo (<u>6b</u>) adducts by column chromatography on a silica gel column using CHCl₃ : diethyl ether = 7:3 as an eluent. ¹H-NMR (CDCl₃): **Sh** $\delta = 0.75$ (s,3H), 0.90 (s,3H), 0.94 (s,9H), 1.06 (s,3H), 1.40-1.78 (m,6H), 1.79 (bs,1H), 1.84 (d,1H), 1.96 (s,3H), 2.41 (dd,1H), 2.85 (d,1H), 2.91 (bs,1H), 2.96 (bs,1H), 3.25(d,1H), 3.39 (d,1H), 4.75 (d,1H), 5.78-5.83 (m, 1H), 5.87 (bs,1H), 6.34-6.39 (m, 1H). **6b** $\delta = 0.77$ (s,3H), 0.91 (s,12H), 1.08 (s,3H), 1.05-1.25 (m, 2H), 1.38-1.71 (m,4H), 1.75-1.85 (m,2H), 1.87 (s,3H), 2.79-2.94 (m,3H), 3.25-3.33 (m,3H), 4.81 (d,1H), 5.59 (bs, 1H), 6.05-6.12 (m,1H), 6.40-6.50 (m,1H). Found: **5b** C 71.65, H 9.70, N 3.12 %; <u>6b</u> C 72.03, H 9.22, N 3.16 % Calc. for C₂₅H₃₉NO₄ : C 71.91, H 9.41, N 3.35 % m.p. : **5b** = 188 ± 2°C; <u>6b</u> = 156 ± 2°C. **5b** : $[\alpha]_D^{24}$ (c = 6.4 x 10⁻¹, MeOH) = +32.8 ± 0.5 **6b** : $[\alpha]_D^{24}$ (c = 7.9 x 10⁻¹, MeOH) = -61.4 ± 0.5

Characterization and identification of adducts 4b, 5b, 6b, 7b.

* Methyl 2-exo-acetamidobicyclo[2.2.1]hept-5-ene-2-endo-carboxylate (12+13) and methyl 2-endo-acetamidobicyclo[2.2.1]hept-5-ene-2-exo-carboxylate (14+15).

These products were prepared and identified as described in the literature^{2b,3b}.

* Transesterification of $(\underline{12}+\underline{13})$ and $(\underline{14}+\underline{15})$ to give $(\underline{4b}+\underline{5b})$ and $(\underline{6b}+\underline{7b})$.

To an anhydrous toluene solution (15 ml) of (-)-cis-3-hydroxy isobornyl neopentyl ether (2b) (600 mg, 2.5 mmol) kept in an inert atmosphere, a solution of AlMe₃ (1.37 ml solution 2N in hexanes, 2.75 mmol) was added dropwise at 0°C. After stirring at this temperature for 30 min., the ice-bath was removed and a solution of methyl 2-exo-acetamidobicyclo[2.2.1]hept-5-ene-2-endo-carboxylate (12+13) [or the corresponding exo adducts (14+15)] (785 mg, 3.75 mmol) in anhydrous toluene (5 ml) was added. The reaction was refluxed for 4 days. After cooling, the reaction mixture was treated with Na₂CO₃.10H₂O (500 mg), filtered and the solvent evaporated under reduced pressure. The corresponding products 4b + 5b [or 6b + 7b] were purified by column chromatography using CHCl₃ : diethyl ether = 7:3 as an eluent.

Found: 4b+5b C 71.71, H 9.58, N 3.20%; 6b+7b C 72.10, H 9.50, N 3.12% Calc. for C₂₅H₃₉NO₄: C 71.91, H 9.41, N 3.35%

Determination of the absolute configuration of 5b and 6b.

* (-)-Menthyl 2-exo-acetamidobicyblo[2.2.1]hept-5-ene-2-endo-carboxylate (<u>4a+5a</u>) and (-)menthyl 2-endo-acetamidobicyblo[2.2.1]hept-5-ene-2-exo-carboxylate (<u>6a+7a</u>).

Mixtures of 4a + 5a and 6a + 7a of known stereochemical composition were prepared from (-)-menthyl N-acetyl- α , β ,-didehydroalaninate (3a) and identified following the procedure described in the literature^{3b}.

 2-exo-acetamidobicyclo[2.2.1]hept-5-ene-2-endo-carboxylic acid (8+2) from 4a + 5a and 2-endo-acetamido-bicyclo[2.2.1]hept-5-ene-2-exo-carboxylic acid (10+11) from 6a + 7a.

These reactions were performed following the method previously described^{3b}.

* Methyl 2-exo-acetamidobicyclo[2.2.1]hept-5-ene-2-endo-carboxylate (<u>12+13</u>) from (<u>4a+5a</u>).

To a solution of the mixture of 2-exo-acetamidobicyclo[2.2.1]hept-5-ene-2-endo-carboxylic acid ($\underline{8+9}$) coming from ($\underline{4a+5a}$) (195 mg, 1 mmol) in methanol, an ethereal solution of diazomethane is added until the yellow colour remains. The solution is stirred until the reaction is completed (monitored by TLC). The excess of diazomethane is destroyed with CaCl₂, the solution is filtered and the solvent eliminated under reduced pressure to afford quantitatively the mixture of methyl esters ($\underline{12+13}$).

This mixture was identified by ¹H-NMR, which was in good agreement with the data described in the literature^{2b}, and were analyzed by ¹H-NMR in CD₃CN in the presence of Eu(tfc)₃ (L:S molar relationship = 0.85).

<u>12</u> (1S,2R) : NHCO<u>CH</u>₃ (5.50 ppm) ; CO₂CH₃ (4.935 ppm). <u>13</u> (1R,2S) : NHCO<u>CH</u>₃ (5.295 ppm) ; CO₂CH₃ (5.07 ppm).

* Methyl 2-endo-acetamidobicyclo[2.2.1]hept-5-ene-2-exo-carboxylate (<u>14+15</u>) from (<u>6a+7a</u>).

The mixture was quantitatively prepared from the mixture (<u>10+11</u>), coming from (<u>5a+7a</u>), following the above described procedure. The mixture was analized by ¹H-NMR in CD₃CN in the presence of Eu(tfc)₃ (L:S molar relationship = 0.85)

 14
 (1S,2S) : NHCOCH3
 (5.05 ppm);
 CO2CH3
 (4.75 ppm).

 15
 (1R,2R) : NHCOCH3
 (5.30 ppm);
 CO2CH3
 (4.80 ppm).

* (1R,2S,4R)-2-exo-acetamidobicyclo[2.2.1]hept-5-ene-2-endo-carboxylic acid (2).

To a solution of (1R,4R)-2-exo-acetamidobicyclo[2.2.1]hept-5-ene-2-endo-carboxylate of (-)-cis-3hydroxy isobornyl neopentyl ether (<u>5b</u>) (417 mg, 1 mmol) in EtOH (10 ml) was added 7 ml of a solution of KOH 3N in EtOH and the reaction mixture was refluxed for 2 days. The resulting reaction was cooled and the solvent was evaporated. The residue was diluted in water (30 ml) and washed with diethyl ether (3 x 15 ml).

The chiral auxiliary is recovered in 85% yield after washing the organic layer with a solution of NaOH 3N, drying with Na₂SO₄ and evaporation of the solvent.

The aqueous layer was acidified at pH=3 and the white precipitate filtered off. The filtrate was extracted with diethyl ether (20 ml) in a Soxhlet apparatus for 4 days. After extraction, the organic layer was dried over Na₂SO₄ and evaporated under vacuum to afford 120 mg of $\frac{9}{2}$ (yield: 60%).

¹H-NMR and IR data are in agreement with those described in the literature^{3b}.

Found: <u>9</u> C 61.35, H 6.89, N 7.22 % Calc. for C₁₀H₁₃NO₃: C 61.53, H 6.71, N 7.17 % m.p. = $221 \pm 2^{\circ}$ C (dec.) [α] $_{D}^{24}$ (c = 11.4 x 10⁻¹, MeOH) = +156.0 ± 0.5

* (15,25,45)-2-endo-acetamidobicyclo[2.2.1]hept-5-ene-2-exo-carboxylic acid (10).

The reaction was performed using the same conditions described for <u>9</u>. Starting from <u>6b</u> (417 mg, 1 mmol), 1 day for reaction, 2 days for continuous extraction, 140 mg of <u>10</u> were obtained (yield: 72%).

¹H-NMR and IR data are in agreement with those described in the literature^{3b}.

Found: <u>10</u> C 61.35, H 6.51, N 7.40 % Calc. for $C_{10}H_{13}NO_3$: C 61.53, H 6.71, N 7.17 % m.p. = $224 \pm 2^{\circ}$ C (dec.) [α] $_{D}^{24}$ (c = 12×10^{-1} , MeOH) = $-106.2 \pm 0.5^{\circ}$

* Methyl (1R,2S,4R)-2-exo-acetamidobicyclo[2.2.1]hept-5-ene-2-endo-carboxylate (13).

The product was quantitatively obtained from 2, following the above described procedure of methylation with diazomethane.

¹H-NMR data are in good agreement with those described in the literature^{2b}.

Absolute configuration was assigned by comparing the ¹H-NMR analysis in the presence of Eu(tfc)₃ (L:S molar relationship = 0.85, CD₃CN) with the above described data for <u>12</u> and <u>13</u>.

Found: <u>13</u> C 63.09, H 7.20, N 6.55 % Calc. for C₁₁H₁₅NO₃: C 63.14, H 7.23, N 6.69 % m.p. = $123 \pm 2^{\circ}$ C. [α]D²⁴ (c = 12.75 x 10⁻¹, MeOH) = +72.5 \pm 0.5

* Methyl (1S,2S,4S)-2-endo-acetamidobicyclo[2.2.1]hept-5-ene-2-exo-carboxylate (14).

The product was quantitatively obtained from <u>10</u>, following the above described procedure of methylation with diazomethane.

¹H-NMR data are in good agreement with those described in the literature^{2b}.

Absolute configuration was assigned by comparing the ¹H-NMR analysis in the presence of Eu(tfc)₃ (L:S molar relationship = 0.85, CD₃CN) with the above described data for <u>14</u> and <u>15</u>.

Found: <u>14</u> C 63.15, H 7.30, N 6.89 % Calc. for C₁₁H₁₅NO₃: C 63.14, H 7.23, N 6.69 % m.p. = $132 \pm 2^{\circ}$ C. [α] $_{D}^{24}$ (c = 17.90 x 10⁻¹, MeOH) = -97.3 ± 0.5

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